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ESTIMATE OF THE PARTICULATE HAZARD

IN UNDERGROUND EXPLOSION OF 1 KT U-235 BOMB

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1. In connection with the underground detonation of an atomic bomb, the inhalation of a high specific activity dust particle and the subsequent lodgment of this particle in the lung has been suggested as a possible source of radiological hazard. In order to evaluate this hazard it is necessary to determine the maximum specific activity which such a particle might have. The most probable size of a particle which might remain lodged in the lung is about one micron. Larger particles tend to be trapped in the nasal passages and subsequently expelled. Furthermore, large particles (100 microns or greater) will not remain airborne for any appreciable distance of travel. Nevertheless, in order to evaluate the possible hazard, it is desirable to calculate the maximum specific activity per particle which might be obtained on particles up to 100 microns.
2. The volume of earth mixed with the fission products will be greater than the total volume of earth included in the fireball and less than the total volume of earth thrown up from the crater. Since mixing within the fireball is certain to be complete, the fireball volume will give the absolute maximum specific activity per particle while the crater volume will give a more reasonable average specific activity. These two volumes are therefore used to cover the range of specific

CLASSIFICATION ~~CONFIDENTIAL~~ which may be encountered.  
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*Jose Diaz 4/22/81*  
REVIEWED BY DATE  
*Wilbur A. Strouser 4/22/81*  
By: W. Tench 7/30/87

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3. Calculation of Specific Activity per Particle.

$$\text{Vol of 1 micron } (\mu) \text{ particle} = 4 \times 10^{-12} \text{ cm}^3$$

$$1 \text{ yd}^3 = 8 \times 10^5 \text{ cm}^3$$

$$\therefore 1 \text{ yd}^3 = 2 \times 10^{17} 1/\mu\text{ particles}$$

$$\text{Total } \beta \text{ activity at 24 hr} = 1.3 \times 10^7 \text{ curies}$$

$$\text{Sr}^{89} \text{ (53 day half-life)} = 3 \times 10^4 \text{ curies}$$

$$\text{Sr}^{90} - \gamma^{90} \text{ (25 yr half-life)} = 5 \times 10^3 \text{ curies}$$

Radius	<u>Fireball</u> = 15 feet	<u>Crater</u> 200 feet
Depth	(15) "	120 "
Vol of earth	500 yd <sup>3</sup>	1.5 x 10 <sup>5</sup> yd <sup>3</sup>
Total 1/ $\mu$ particles	10 <sup>20</sup>	3 x 10 <sup>22</sup>
Total $\beta$ activity per 1/ $\mu$ particle	1.3 x 10 <sup>-13</sup> curies	4 x 10 <sup>-16</sup> curies
Sr <sup>89</sup> activity per 1/ $\mu$ particle	3 x 10 <sup>-16</sup> curies	1 x 10 <sup>-18</sup> curies
Sr <sup>90</sup> - $\gamma^{90}$ " " " "	5 x 10 <sup>-17</sup> curies	2 x 10 <sup>-19</sup> curies

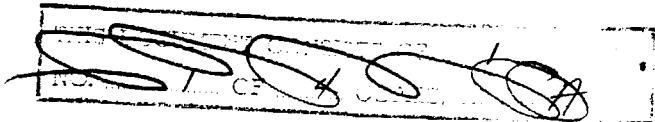
For 10/ $\mu$  particle activity will be 10<sup>3</sup> times greater than for 1 micron particle  
 For 100/ $\mu$  particle activity will be 10<sup>6</sup> times greater than for 1 micron particle

CALCULATION OF HAZARDS FROM SURFACE OR UNDERGROUND DETONATIONS

Assumptions for Worst Possible Situation:

- a. All radioactive material stays in cloud and none is deposited in or around crater.
- b. Cloud moves along ground and does not lift.
- c. Cloud spread and rise minimized.

1 KT U-235 Bomb; Wind 10 mi/hr or less; inhalation rate  $10 \text{ m}^3 / 8 \text{ hrs}$  ( $20 \text{ l/min.}$ )



Time hrs	Total $\gamma$ Curies	Total $\beta$ Curies	Sr <sup>89</sup> Curies 0.1% @ 12 hr (53 d)	Sr <sup>90</sup> + Y 90 Curies 0.2% @ 26 d (25 y)	Cloud			Transit Time min.	Maximum Distance Traveled mi.	Amount Inhaled Liter	Concentration (Curies/liter)			
					dia. mi.	thickness mi.	Volume $\text{mi.}^3$				$\beta$	Sr <sup>89</sup>	Sr <sup>90</sup> Y 90 $\gamma$	
1	$30 \times 10^7$	$60 \times 10^7$	$3 \times 10^4$	$5 \times 10^{3.2}$										
3	$8 \times 10^7$	$16 \times 10^7$	$3 \times 10^4$	$5 \times 10^{3.2}$	2	1/5	0.6	$3 \times 10^{12}$	12	30	240	$5 \times 10^{-5}$	$1 \times 10^{-8}$	$2 \times 10^{-9}$ $3 \times 10^{-5}$
6	$3.5 \times 10^7$	$7 \times 10^7$	$3 \times 10^4$	$5 \times 10^{3.2}$	3	1/5	1.4	$6 \times 10^{12}$	18	60	360	$1 \times 10^{-5}$	$5 \times 10^{-9}$	$1 \times 10^{-9}$ $5 \times 10^{-6}$
12	$1.5 \times 10^7$	$3 \times 10^7$	$3 \times 10^4$	$5 \times 10^{3.2}$	6	2/5	11	$50 \times 10^{12}$	36	120	720	$6 \times 10^{-7}$	$6 \times 10^{-10}$	$1 \times 10^{-10}$ $3 \times 10^{-7}$
24	$0.7 \times 10^7$	$1.3 \times 10^7$	$3 \times 10^4$	$5 \times 10^{3.2}$	10	1/2	40	$200 \times 10^{12}$	60	240	1200	$7 \times 10^{-8}$	$2 \times 10^{-10}$	$3 \times 10^{-11}$ $4 \times 10^{-7}$

$\gamma$ -intensity (r/hr)  $I_c = 3 \times 10^5 \times \text{Conc in } \gamma \text{ curies/liter}$

Tolerances - Chalk River Conference

Time hrs	Total $\gamma$ Curies	INHALATION		External Dose		1% of inhaled Sr <sup>89</sup> - <sup>90</sup> is absorbed Same for Sr <sup>89</sup>
		Inhaled Absorbed	Sr <sup>89</sup> $\mu$ curies Inhaled Absorbed	Sr <sup>90</sup> - Y 90 $\mu$ curies Inhaled Absorbed	Ic r/hr	
3	$1.2 \times 10^{-2}$	(10%) $1.2 \times 10^{-3}$	2.4 0.02	.0.5 0.005, .003	9 2	
6	$4.2 \times 10^{-3}$	$3.6 \times 10^{-4}$	1.8 0.018	.0.36 0.003, .002	1.5 0.45	
12	$4.3 \times 10^{-4}$	$4.3 \times 10^{-5}$	0.4 0.004	.0.07 0.0007, .0007	0.9 0.054	
24	$8 \times 10^{-5}$	$8 \times 10^{-6}$	0.24 0.002	.0.04 0.0004, .0004	0.1 0.012	

1% of inhaled Sr <sup>90</sup> - <sup>90</sup> is absorbed

Same for Sr <sup>89</sup>

$\checkmark$  Sr <sup>89</sup> - 1  $\mu$ c absorbed in body

$\checkmark$  Sr <sup>90</sup> - 0.5  $\mu$ c absorbed in body